


Pyrimido(4,5-g)quinazoline derivatives with anti-tumour activity.

Patent Number: ☐ EP0541404
Publication date: 1993-05-12
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Requested Patent: ☐ US5639881
Application Number: EP19920310231 19921109
Priority Number(s): US19910763375 19911108
IPC Classification: A61K31/505 ; C07D487/04
EC Classification: C07D487/04
Equivalents: CA2082297, ☐ JP7309873, MX9206438

Abstract

Pyrimido-4,5-g-quinazoline quinone derivatives were synthesized as anthraquinone-like reductive alkylating agents. Like many naturally-occurring antibiotics, these quinone derivatives are designed to afford an alkylating quinone methide species upon reduction and leaving group elimination. Kinetic studies of pyrimido-4,5-g-quinazoline hydroquinones provided evidence of quinone methide intermediates able to trap nucleophiles (alkylation) and protons. The rate of quinone methide formation is determined by the hydroquinone free energy. Thus, a linear free energy relationship for quinone methide formation was obtained by plotting rates of quinone methide formation, as the log, versus the quinone reduction potential. The pyrimido-4,5-g-quinazoline quinone methides fall on this free energy plot, showing that these species are formed by the same mechanism as the other structurally-diverse quinone methides previously studied in this research group. A drawback of many quinone antibiotics, particularly the anthracyclines, is the formation of toxic oxygen species by quinone/hydroquinone cycling. In the present invention pyrimido-4,5-g-quinazoline hydroquinones are found to be relatively stable toward oxygen, and thus cause little oxygen toxicity. Antitumor screening revealed that the disclosed pyrimido-4,5-g-quinazoline dione and tetione derivatives possess excellent inhibitory activity against selected human cancer cell lines. 

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— Please see exhibit 14 —